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Modern Chemistry Chapter 10. STUDY. PLAY. kinetic-molecular theory, based on idea that particles of matter are always in motion. ideal gas, hypothetical gas that fulfills 5 assumptions. assumption 1. gases consist of large numbers of tiny particles that are far apart relative to their size.

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CHAPTER 10 REVIEW States of Matter SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. a Liquids possess all the following properties except (a) relatively low density. (c) relative incompressibility. (b) the ability to diffuse. (d) the ability to change to a gas. 2. a.

10 States of Matter - Ms. Agostine's Chemistry Page

Modern Chemistry; Holt, Rinehart, and Winston Blue Book. Terms in this set (20) Kinetic Theory of Liquids. 1. Particles in a liquid are in constant motion, but have less mobility than gas particles. 2. Liquids are closer together than particles in a gas. 3. Attractive forces between particles of a liquid are more effective than those between ...

Section 10.2: States of Matter - Liquids - Study Guide ...

General Chemistry: Principles and Modern Applications (10th Edition) Petrucci, Ralph H.; Herring, F. Geoffrey; Madura, Jeffrey D.; Bissonnette, Carey Publisher Pearson Prentice Hal ISBN 978-0-13206-452-1

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Modern Chemistry - CHAPTER 10 HOMEWORK 10-8 (pp. 322-325) VOCABULARY Complete each sentence. 1. The partial pressure of a gas is _____. 2. Dalton's law of partial pressures states that ____

HOMEWORK 10-1

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