

# Chemical Reaction Test Answers

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## Chemical Reaction Test Answers

We thoroughly check each answer to a question to provide you with the most correct answers. Found a mistake? Let us know about it through the REPORT button at the bottom of the page. Click to rate this post! [Total: 0 Average: 0] The chemical reactions that occur inside the human body are extremely complex. ... Chapter 11 Chemical Reactions Test Answer Key Read More »

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There are many different types of chemical reactions. There are single and double displacement reactions, combustion reactions, decomposition reactions, and synthesis reactions. See if you can

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identify the type of reaction in this ten question chemical reaction classification practice test. Answers appear after the final question.

## **Chemical Reaction Classification Practice Test**

Select the best answer and then transfer your answers to the SCANTRON provided. 1. When a piece of magnesium metal is placed into a test tube of hydrochloric acid, bubbles form. What type of reaction is this? (a) synthesis (c) single displacement (b) decomposition (d) double displacement 2. When sodium bicarbonate is heated in a test tube, water vapour, carbon dioxide

## **Chemical Reactions Practice Test Answers**

Answer: In decomposition reactions, a single reactant is broken down into two or more products. Question:  $\text{Zn} + 2\text{HCl} \rightarrow \text{H}_2 + \text{ZnCl}_2$  Answer: Also known as substitution or single replacement, a single displacement reaction consists of a single element replacing one of the elements in a compound. Question:  $\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$

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The balls next to the correct answers will turn green so you can see the right answers.

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Answers 1.  $1 \text{ SnO}_2 + 2 \text{ H}_2 \rightarrow 1 \text{ Sn} + 2 \text{ H}_2\text{O}$  2.  $3 \text{ KOH} + 1 \text{ H}_3\text{PO}_4 \rightarrow 1 \text{ K}_3\text{PO}_4 + 3 \text{ H}_2\text{O}$  3.  $2 \text{ KNO}_3 + 1 \text{ H}_2\text{CO}_3 \rightarrow 1 \text{ K}_2\text{CO}_3 + 2 \text{ HNO}_3$  4.  $1 \text{ Na}_3\text{PO}_4 + 3 \text{ HCl} \rightarrow 3 \text{ NaCl} + 1 \text{ H}_3\text{PO}_4$  5.  $1 \text{ TiCl}_4 + 2 \text{ H}_2\text{O} \rightarrow 1 \text{ TiO}_2 + 4 \text{ HCl}$  6.  $1 \text{ C}_2\text{H}_6\text{O} + 3 \text{ O}_2 \rightarrow 2 \text{ CO}_2 + 3 \text{ H}_2\text{O}$  7.  $2 \text{ Fe} + 6 \text{ HC}_2\text{H}_3\text{O}_2 \rightarrow 2 \text{ Fe}(\text{C}_2\text{H}_3\text{O}_2)_3 + 3 \text{ H}_2$  8.  $4 \text{ NH}_3 + 5 \text{ O}_2 \rightarrow 4 \text{ NO} + 6 \text{ H}_2\text{O}$  9.  $1 \text{ B}_2\text{Br}_6 + 6 \text{ HNO}_3 \rightarrow 2 \text{ B}(\text{NO}_3)_3 + 6 \text{ HBr}$  10.  $4 \text{ NH}_4\text{OH} + 1 \text{ K}_2(\text{SO}_4)_2 \cdot 12 \text{ H}_2\text{O}$

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→ 1 Al(OH)<sub>3</sub> + 2 (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> ...

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Use the graph below to answer questions 24 and 25. \_\_\_\_ 24. Reading Graphs This graph shows the change in energy during a chemical reaction. Which has more energy? a. the reactants b. the products c. The reactants and the products have the same amount of energy. \_\_\_\_ 25. Applying Concepts The chemical reaction described by this graph is a ...

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