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Chapter 12 Stoichiometry 127 SECTION 12.1 THE ARITHMETIC OF EQUATIONS (pages 353-358) This section explains how to calculate the amount of reactants required or product formed in a nonchemical process. It teaches you how to interpret chemical equations in terms of interacting moles, representative particles, masses, and gas volume at STP.

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What is stoichiometry? The study of quantitative relationships between amounts of reactants used and products formed by a chemical reaction In other words: you must know how much reactant to use in order to get a desired amount of product!

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